

Homework #6**Chapter 26**

1. McQuarrie and Simon Problems: 1,3,10,13,22,25,29,38,58,61,64
2. Consider the reaction below.



Predict what happens at equilibrium if:

- a. The temperature is raised.
- b. The pressure on the system is increased.
- c. An inert gas is added to the system at constant pressure
- d. A catalyst is added to the system.

For Presentation (Nov 27th):

Group E: 26-3

Group F: 26-38 (Just show how to set-up: equations, etc and not all the minute details)

Group G: 26-61