

# ROTD 3/2/06

1) Resonance is really useful when trying to decide the relative stability of molecules. The more resonance structures you can draw for a given molecule, the more stable it is.

2)  $\Delta G = -RT \ln (K_{eq})$ . If  $K_{eq} > 1$ , then  $\Delta G$  will be negative (favors products). If  $K_{eq} < 1$ , the  $\Delta G$  will be positive (favors reactants).  $pK_a$  helps figure out  $K_{eq}$  quickly.

3)  $pK_a$  tables are your friend. But it helps to memorize a few  $pK_a$  values.