## **Problem Set 1**

Following are the group assignments for group homework #1. The goals of this activity are to challenge you to work problems that are more complex and to practice working effectively in a group setting with people that you do not necessarily know well. The person with an 'O' after their name will be the organizer of the group and will be required to arrange meeting times and places to complete the assignment. The person with a 'W' after their name will be the person who prepares the final written assignment to be turned in for a grade. Groups will be re-assigned and roles will be changed for each set of homework, so that you have the opportunity to work with a variety of different people and in different roles throughout the semester.

On the next page is the 'grading' scale for each member of your group. After you have completed the homework assignment, please 'grade' the performance of each group member, including yourself, and turn in separately from the group homework for confidentiality purposes. A person who does not show up for the group meetings or does not participate in the problem solving process should receive an 'F' for the group grade. A person who contributes, asks questions, and helps the other group members to understand the problem and answer deserves an 'A'. A person who requires a little extra help and needs another group member to explain a topic **does not** deserve an 'F'. Their lack of understanding will help you as well because you will have to clearly explain the topic to him/her. If you choose to not turn in grades for your group members, I will assume 'average' participation for the entire group.

Submit only one set of answers per group. Please work with your assigned partners to prepare your final draft of answers. You may find it useful to xerox copies of the answer set, so that each group member may have it for reference. Please do not consult with anyone other than people in your group; you may consult me for question clarification only. You may use your textbooks and any other written references from the library. Do not copy anyone else's work. The honor code pledge is printed on the top of the "grading" scale page. By turning this page in, you are acknowledging that you have read and understand the privileges and responsibilities that this code bestows.

		1	1
Alex Burum	Andrew Bryan (O)	Kristen Burson (W)	Kyle Carlson
Danielle Forstner (O)	Erin Ge	Bryce Gode	Nissa Hannemann
Jeffrey Rock	Doug Schroeder	Bonginkosi Sibiya (O)	Jill Verchota
Ellen Sauter	Sunny Sonnabend (W)	Michael Stangler	Zach Alwine (W)
Ben Treichel (W)	Luis Valle	Krista Cruse	Bridget Hoesley (O)
Holly Cooper	Dan Freeman (O)	Maari Hanson	Ben Levy
Steve Howard (O)	Scott Kyser (W)	Krystal Long (W)	Brittany Murphy (O)
Zach Walgenbach (W)	Alex Burleigh	Chris Ditlevson	Eric Kraska
Michael Butterworth	Nate Erickson	Jennifer Krantz ( <b>O</b> )	Raychal Zupancich (W)
Lauren Hom	Fue Vang	Alyssa Brooks	Brian Castle
Steph Lewis (O)	Chris Lund	Emily Pelton	Amy Waldner
Kathryn Pesch	Matthew Royer (O)	Stephanie Soiseth	Kelly Rozenboom
Nick Malm (W)	Eric Nelson (O)	Rachel Roberg (W)	Danielle Burras (O)
Rachael Chaska	Micah Deitz	Douglas Durand (O)	Sarah Erickson (W)
Matthew Hoke	Timothy Lamanna	Jenna Paulsen	Rachel Poppy
Leah Swanson	Brianna Vaa (O)	Morgan Wells	Trevor Wittwer
David Wold (O)	Conner Ziegler	Nate Bower (O)	Arie DeGrio
Ryan Casper	Mackenzie Consoer	Rachel Elvebak (W)	Jenna Kesty (O)
Aaron Insley (W)	Brent McConahey (W)	Kristen Oldenburg	Eric Anderson (W)
Marcus Perry	Aaron Roessler	_	
David Wray (W)	Emily Barnard (O)		
Sarah Duncan (O)	Justin Hahn		
Chris Leonard	Matija Novakovic (W)		
Kari Maffit	Chad Olson		

## This is due Friday, September 16, at the beginning of class.

## **Peer Evaluation Grading Sheet**

On my honor, I pledge that I have not given, received, nor tolerated others' use of unauthorized aid in completing this work.

	 А	В	С	D	F
Group member #1					
	 А	В	С	D	F
Group member #2					
	 А	В	С	D	F
Group member #3					
	 А	В	С	D	F
Group member #4					
	 А	В	С	D	F
Group member #5					

Comments:

1) (20 pts) A molecule with a molecular formula of  $C_{10}H_{16}O$  was isolated from the essential oil of *Artemisia vulgaris*. The waxy solid was found to have an optical rotation of  $[\alpha]^{25}D = +44.1^{\circ}$  (*c*=10, EtOH). It gave a positive qualitative functional group test for DNP, but negative for the ignition, iodoform and chromic acid tests. It does not discolor bromine solutions. On the following pages are several spectra of the compound (IR, <sup>1</sup>H NMR, <sup>13</sup>C NMR, DEPT, Hetcor).

Propose a structure and assign the resonances in the IR,  ${}^{1}$ H NMR, and  ${}^{13}$ C NMR. (Do the best you can and be sure to provide some reasoning behind the assignment.)











2) (15 pts) Moclobernide (1, marketed as Manerix by Roche) is an inhibitor of the enzyme monoamine oxidase (MAO). Clinically, this means that is can be used to treat depression. Neurotransmitters that contain a primary amine functional group, such as seratonin (2), norepinepherine (3), and dopamine (4), appear to play a major role in mood and emotion. Neurotransmitters are released from one neuron (nerve cell) into the synaptic cleft (the area between an axon and dendrite from an adjacent neuron). These molecules then find specific places (called receptors) on the surface of the dendrite and cause the neuron associated with that dendrite to "fire." This causes the transmission of information from one neuron to another. Once the neuron fires, the need for the neurotransmitter diminishes, so the synaptic cleft must be cleared of the signaling molecules. This can happen either by "reuptake" (absorption of the molecule back into the axon from which it came) or it can be destroyed in the synaptic cleft. The molecules 2, 3, and 4 are cleared through a degradation process mediated by MAO. Depression often results from a lack or imbalance of these neurotransmitters. By inhibiting the mechanism that destroys them, the concentration of these molecules can be increased, and the clinical signs of depression are alleviated.

The synthesis of moclobemide starts with mopholine (5) and ends with the acylation of 6. Propose a synthetic route for the conversion of 5 to 6. You may use any inorganic reagents, but organic reagents must contain 2 or fewer carbons.



3) (15 pts) Provide a mechanism, using curved arrow notation, that explains the following reaction:



4) (15 pts) Morphine (7) and codeine (8) are isolated from opium poppies. Morphine is a potent analgesic (pain reliever). Codeine is not nearly as good an analgesic, but is a good antitussive (cough suppressant). It is thought that codeine is not an analgesic at all, but rather is converted to morphine by the action of an enzyme. Without knowing what the enzyme is, what is the mechanism through which this enzyme acts (*i.e.* use curved arrow notation)? It might help if you use H-A, B<sup>-</sup>, and Nu<sup>-</sup> as general "reagents."



5) (15 pts) Sugars that contain 6-membered rings are important biological molecules. Although glucose (9) is the most common example used, there are many sugars that play important roles. In many biochemical reactions that involve sugars, the carbon bearing two single bonds to oxygen (*i.e.* the carbon labeled a) is the reactive center. To interrupt the enzymes that carry out these various transformations, chemists can make molecules that look like the substrate for the enzyme, but which cannot actually do the chemistry of the native sugar substrate, by removing the oxygen found in the ring of the sugars. In essence, the enzyme is tricked into thinking it is bound to the right molecule but gets stuck when it cannot carry out the right chemical transformation. This ties the enzyme up and "removes" it from the chemical equilibria in the cell.

Starting from cyclohexylbromide (10), show how one might make the sugar mimic *trans*-1,2-dihydroxycyclohexane (11).



6) (20 pts) Given what you have learned about the reaction of ethers with Brønsted acids, propose a what product(s) will be formed and provide a mechanism, using curved arrow notation, for the following reaction:



## Extra Credit: (10 pts)

Meperidine (12) was discovered in Germany in 1937. It is only 1/8<sup>th</sup> as potent as morphine in pain relief, but is much shorter acting and is more quickly metabolized. Consequently, it was thought that this drug might be a good substitute for morphine without having the liability or tolerance or addition. Unfortunately, this is not the case. It is, however, still used as an obstetric and anesthetic premedication. As an obstetric analgesic, it appears to depress the respiration of the fetus less than morphine and does not cause constipation like morphine (much to the relief of the patient). The following is a "road map" for the synthesis of meperidine. Fill in the missing reagents/products for the scheme. You may need to consult chapters in your book that have not yet been covered, but you actually know everything you need to figure it out.

